

1,13-Dimethyl-6H,7H,8H-chromeno-[3',4':5,6]pyrano[3,2-c]chromene-6,8-dione

M. Iqbal Choudhary,^a Alain Meli Lannang,^b Shazia Anjum,^{a*} Hoong-Kun Fun,^{c,‡} David Lontsi^b and Jean Gustave Tangmouo^b

^aHEJ Research Institute of Chemistry, International Centre for Chemical Sciences, University of Karachi, Karachi 75270, Pakistan,
^bDepartment of Organic Chemistry, Faculty of Science, University of Yaounde I, PO Box 812, Cameroon, and ^cX-ray Crystallography Unit, School of Physics, Universiti Sains Malaysia, 11800 USM, Penang, Malaysia

* Additional correspondence author, email:
 hkfun@usm.my

Correspondence e-mail:
 anjumshazia@yahoo.com

Key indicators

Single-crystal X-ray study
 T = 293 K
 Mean $\sigma(C-C) = 0.004 \text{ \AA}$
 R factor = 0.047
 wR factor = 0.136
 Data-to-parameter ratio = 11.4

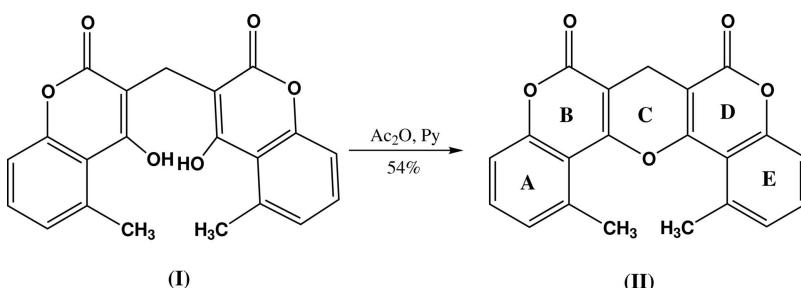
For details of how these key indicators were automatically derived from the article, see
<http://journals.iucr.org/e>.

The title compound, $C_{21}H_{14}O_5$, is a pentacyclic chromene derivative. The dihedral angle between the two outer benzene rings is $30.3(1)^\circ$. The pyran and one of the pyranone rings adopt twist-boat conformations. The crystal packing is stabilized by $C-H\cdots O$, $C-H\cdots\pi$ and $\pi\cdots\pi$ interactions

Received 2 December 2005
 Accepted 5 December 2005
 Online 10 December 2005

Comment

Diospyros canaliculata De Wildeman (Ebenaceae) is a tree up to 15–20 m tall (Letouzey & White, 1970) found throughout the forest zone of West and Central Africa (Jeffreys *et al.*, 1983). The bark extract is poisonous and causes blistering and gangrene around wounds, and is a common ingredient of the Guere (Ivory Coast) arrow poisons (Irvine, 1961). Previous investigation of the stem bark of this plant led to the isolation of gerberinol, (I) (Sengupta *et al.*, 1985; Paknikar *et al.*, 1996; Tangmouo *et al.*, 2005). In the present work, we have undertaken the X-ray structure determination of the title compound, (II) (Sengupta *et al.*, 1985), derived from (I).



The bond lengths in compound (II) show normal values (Allen *et al.*, 1987). The molecule is non-planar, with the outer benzene rings *A* and *E* forming a dihedral angle of $30.3(1)^\circ$. The pyranone ring *B* is planar, with a weighted average absolute torsion angle (τ) of 4.1° . The pyran ring *C* and pyranone ring *D* are slightly puckered and adopt twist-boat conformations; the Cremer and Pople (1975) puckering parameters Q , θ and φ are $0.185(2) \text{ \AA}$, $82.6(6)^\circ$ and $157.0(8)^\circ$ for ring *C*, and $0.182(2) \text{ \AA}$, $89.6(9)^\circ$ and $30.9(8)^\circ$ for ring *D*, respectively. The $C1-C14B-C14A$ [$128.3(2)^\circ$] and $C13-C13A-C13B$ [$128.30(2)^\circ$] angles are larger than 120° as a result of steric interactions between the methyl groups at the *C1* and *C13* positions.

As shown in Fig. 1, the $C16-H16C\cdots O3$ and $C15-H15C\cdots O3$ hydrogen bonds generate rings of graph-set motif $S(6)$ (Bernstein *et al.*, 1995). The crystal packing is stabilized by $C-H\cdots O$, $C-H\cdots\pi$ (Table 1) and $\pi\cdots\pi$ interactions. The $\pi\cdots\pi$ interactions involve rings *A*, *B* and *E* of inversion-related molecules with $Cg2\cdots Cg2^{ii} = 3.593(3) \text{ \AA}$, $Cg1\cdots Cg1^{iii} = 3.544(3) \text{ \AA}$ and $Cg3\cdots Cg3^{iii} = 3.790(4) \text{ \AA}$ [symmetry codes:

(ii) $1-x, 1-y, 1-z$; (iii) $1-x, 2-y, 1-z$; (iii) $1-x, 1-y, -z$, where $Cg1$, $Cg2$ and $Cg3$ are the centroids of rings A , B and E , respectively. The C—H··· π interaction involves ring A . A view of the molecular packing is shown in Fig. 2.

Experimental

Compound (II) was prepared from gerberinol (I) using the literature method of Sengupta *et al.* (1985). A solution of (I) (15 mg) in pyridine (0.5 ml) and acetic anhydride (0.5 ml) was stirred for 2 d at room temperature. The solvent was evaporated and the resulting solid pink residue was crystallized from $\text{CHCl}_3\text{--CH}_3\text{OH}$ (1:1 v/v) to afford compound (II) as shiny pink crystals (yield 54%, 8 mg; m.p. 523 K).

Crystal data

| | |
|--------------------------------|---|
| $C_{21}H_{14}O_5$ | $Z = 2$ |
| $M_r = 346.32$ | $D_x = 1.495 \text{ Mg m}^{-3}$ |
| Triclinic, $P\bar{1}$ | Mo $K\alpha$ radiation |
| $a = 8.052 (6) \text{ \AA}$ | Cell parameters from 3408 |
| $b = 8.492 (7) \text{ \AA}$ | reflections |
| $c = 12.544 (10) \text{ \AA}$ | $\theta = 1.7\text{--}25.0^\circ$ |
| $\alpha = 91.441 (13)^\circ$ | $\mu = 0.11 \text{ mm}^{-1}$ |
| $\beta = 106.324 (13)^\circ$ | $T = 293 (2) \text{ K}$ |
| $\gamma = 109.489 (13)^\circ$ | Plate, pink |
| $V = 769.2 (11) \text{ \AA}^3$ | $0.24 \times 0.12 \times 0.06 \text{ mm}$ |

Data collection

Siemens SMART CCD area-detector diffractometer
 ω scans
Absorption correction: multi-scan (SADABS; Sheldrick, 1996)
 $T_{\min} = 0.975$, $T_{\max} = 0.994$
6917 measured reflections

2695 independent reflections
1842 reflections with $I > 2\sigma(I)$
 $R_{\text{int}} = 0.036$
 $\theta_{\text{max}} = 25.0^\circ$
 $h = -9 \rightarrow 9$
 $k = -10 \rightarrow 10$
 $l = -14 \rightarrow 14$

Refinement

Refinement on F^2
 $R[F^2 > 2\sigma(F^2)] = 0.047$
 $wR(F^2) = 0.136$
 $S = 1.05$
2695 reflections
237 parameters
H-atom parameters constrained

$$w = 1/[\sigma^2(F_o^2) + (0.0615P)^2 + 0.1079P]$$

where $P = (F_o^2 + 2F_c^2)/3$

$$(\Delta/\sigma)_{\text{max}} = 0.001$$

$$\Delta\rho_{\text{max}} = 0.25 \text{ e \AA}^{-3}$$

$$\Delta\rho_{\text{min}} = -0.19 \text{ e \AA}^{-3}$$

Table 1

Hydrogen-bond geometry (\AA , $^\circ$).

$Cg1$, $Cg2$ and $Cg3$ are the centroids of rings A , B and E , respectively.

| $D-\text{H}\cdots A$ | $D-\text{H}$ | $\text{H}\cdots A$ | $D\cdots A$ | $D-\text{H}\cdots A$ |
|--------------------------------|--------------|--------------------|-------------|----------------------|
| $C2-\text{H}2A\cdots O1^i$ | 0.93 | 2.57 | 3.442 (4) | 155 |
| $C15-\text{H}15C\cdots O3$ | 0.96 | 2.43 | 2.848 (4) | 106 |
| $C16-\text{H}16C\cdots O3$ | 0.96 | 2.37 | 2.937 (4) | 117 |
| $C7-\text{H}7A\cdots Cg1^{ii}$ | 0.97 | 2.84 | 3.583 (4) | 134 |

Symmetry codes: (i) $x+1, y+1, z$; (ii) $-x+1, -y+1, -z+1$.

All H atoms were positioned geometrically and allowed to ride on the parent carbon atoms, with C—H distances in the range 0.93–0.97 \AA and $U_{\text{iso}}(\text{H}) = 1.5U_{\text{eq}}(\text{C})$ for methyl H atoms and $1.2U_{\text{eq}}(\text{C})$ for other H atoms.

Data collection: SMART (Siemens, 1996); cell refinement: SAINT (Siemens, 1996); data reduction: SAINT; program(s) used to solve structure: SHELLXTL (Sheldrick, 1997); program(s) used to refine structure: SHELLXTL; molecular graphics: SHELLXTL; software used to prepare material for publication: SHELLXTL and PLATON (Spek, 2003).

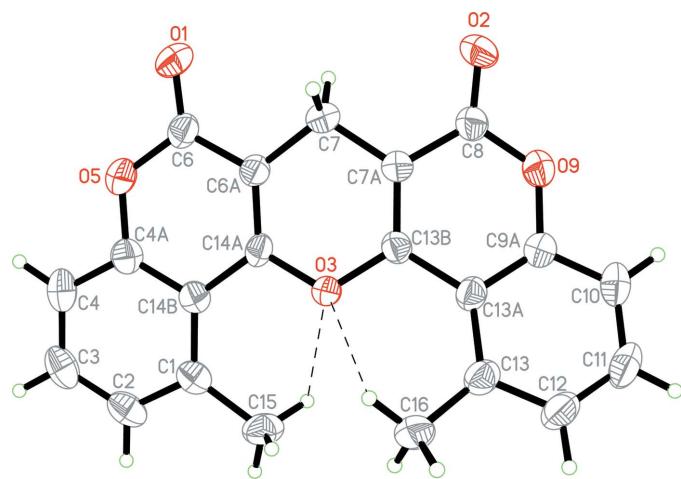


Figure 1

The molecular structure of (II), showing 50% probability displacement ellipsoids and the atom-numbering scheme. Dashed lines indicate hydrogen bonds.

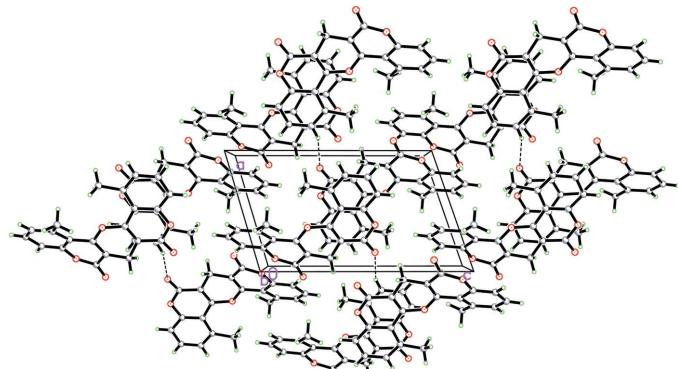


Figure 2

The crystal packing of (II), viewed approximately along the b axis. Dashed lines indicate hydrogen bonds.

AML thanks the TWAS for financially supporting his training in ICCS HEJ RIC, University of Karachi, Pakistan. SA and HKF thank the Malaysian Government and the Universiti Sains Malaysia for Scientific Advancement Grant Allocation (SAGA) No. 304/PFIZIK/653003/A118.

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–S19.
- Bernstein, J., Davis, R. E., Shimoni, L. & Chang, N.-L. (1995). *Angew. Chem. Int. Ed. Engl.* **34**, 1555–1573.
- Cremer, D. & Pople, J. A. (1975). *J. Am. Chem. Soc.* **97**, 1354–1358.
- Irvine, F. R. (1961). *Woody Plants of Ghana*, pp. 576–578. London: Oxford University Press.
- Jeffreys, J. A. D., Muhamad, B. Z., Waterman, P. G. & Zhong, S. M. (1983). *Tetrahedron Lett.* **24**, 1085–1088.
- Letouzey, R. & White, F. (1970). *Flore du Cameroun: Ebénacées*, vol. 11, p. 43. Muséum National d'Histoire Naturelle, 16 rue Buffon, Paris 5, France.
- Paknikar, S. K., Pai, F. K. P., Kirtany, J. K. & Natori, S. (1996). *Phytochemistry*, **41**, 931–933.
- Sengupta, P., Sen, M., Karuri, P., Wenkert, E. & Halls, T. D. J. (1985). *J. Indian Chem. Soc.* **62**, 916–919.
- Sheldrick, G. M. (1996). *SADABS*. University of Göttingen, Germany.

- Sheldrick, G. M. (1997). *SHELXTL*. Version 5.1. Bruker AXS Inc., Madison, Wisconsin, USA.
- Siemens (1996). *SMART* and *SAINT*. Siemens Analytical X-Ray Instruments Inc., Madison, Wisconsin, USA.
- Spek, A. L. (2003). *J. Appl. Cryst.* **36**, 7–13.
- Tangmouo, J. G., Lontsi, D., Ngounou, F. N., Kuete, V., Meli, A. L., Manfouo, R. N., Kamdem, H. W., Tane, P., Penlap, B. V., Sondengam, B. L. & Connolly, J. D. (2005). *Bull. Chem. Soc. Ethiop.* **19**, 81–88.